DEFECT-FLUORITE OXIDES: Ln (Eu AND Gd)-MÖSSBAUER SPECTROSCOPIC STUDY COUPLED WITH NEW DEFECT-CRYSTAL-CHEMISTRY MODEL

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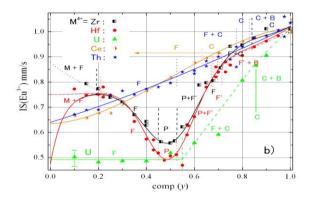
Defect-fluorite (DF) oxides as a generic term of various grossly oxygen-deficit oxygen-vacancy (V_0) type $M_{1-y}Ln_yO_{2-y2}(V_{Oy/2})$ solid solutions (ss) formed between parent fluorite (F) $MO_2(M^{4+} = Zr, Hf, Ce, Th, U, Np and$ Pu etc) and its V₀-ordered superstructure, lanthanide (Ln) and light-Ln C-type $LnO_{1.5}$ (Ln^{3+} = La-Lu, Y, Sc, In *etc*), have been the object of enormous research efforts for over these several decades due to their technological importance in nuclear, ceramic and electrochemical etc areas. Yet, much has remained elusive till now as to their detailed defect (V₀)-related local structure and its correlations with many key defect properties such as oxygen conductivity ($\sigma(ion)$), radiation tolerance, catalytic ability and phase and structural stability etc. To serve this purpose in these years we have been engaged in their combined Ln (Eu and Gd)-Mössbauer and XRD latticeparameter $(a_0(ss))$ studies [1-5], lately extending them to ¹⁷⁰Yb-Mössbauer, EXAFS and Y-MAS-NMR *etc* studies.

One basic controversy of such defect-structure study has been; albeit classified as the same DF oxides, whether those based on larger $M^{4+}s$ (Ce, Th, U, Np, Pu and Am *etc*) which themselves adopt F structure and those based on smaller $M^{4+}s$ (Zr and Hf) which are first stabilized into F structure by $LnO_{1.5}\rightarrow Zr(Hf)O_2$ doping, the both have either mutually 'different or similar' local structure. Our ¹⁵¹Eu-Mössbauer study of $M_{1-y}Ln_yO_{2-y2}$ (M^{4+} = Ce, Th, U, Zr and Hf) (Fig. 1 [1]) has given a clear answer to this controversy: Based on the well-known near inverselyproportional IS(Eu³⁺)-(Eu³⁺-O) bond-length (BL) relationship, these data decisively reveal that:

(i)The parent F-based M^{4+} =Ce and Th have a na $\ddot{v}e$ (but non-random) 'disordered DF-type' local structure in which both BL(Eu³⁺-O)s decrease smoothly and similarly due to the steady oxygen coordination number (CN) decrease about Eu³⁺ from ~7.5 at *y*=0 to 6 at *y*=1.0.

(ii)The stabilized zirconia (SZ) and stabilized hafnia (SH) ($M^{4+}=Zr$ and Hf) the both have a sharp BL(Eu³⁺-O) maximum, that of the SH with the shorter $a_0(ss)$ being even longer than that of the SZ with the longer $a_0(ss)$. Such characteristic IS(Eu³⁺) behavior of SZ and SH is entirely different from that of the above parent-F-based systems, and is almost a first direct experimental evidence that SZs and SHs have basically the same intermediate ordered pyrochlore (P Zr(Hf)₂Eu₂O₇ at *y*=0.50)-based strongly anisotropic local structure extending to either the *y*<0.50 or >0.50 region retaining its short-range (local) P-type ordering, and thus the both have the longest BL(Eu³⁺-O) at *y*=0.50 at which the most highly-ordered and the most-

dilated near-ideally P-type Eu-80²⁻ (CN=8) coordination structure is realized. Making these findings a major incentive, one of the authors (A.N) has recently proposed a new defect-crystal-chemistry (DCC) $a_0(ss)$ model for parent-F-type M⁴⁺=Ce and Th [2-4] which can provide not only a novel possibly unified picture of their non-Vegardianity and non-random defect structure as closely coupled two sides of 'distortion-dilation' phenomenon of C-type LnO_{1.5} but also a new consistent description of their detailed mutually non-randomly coordinated cationanion concentration variations and also of their intriguing $\sigma(ion)$ maximum behavior in low y (<0.10-0.20) range.



We are now trying to extend this DCC $a_0(ss)$ model to the more complex SZs and SHs involving the intermediate P (or δ)-type ordering and hence its-based extra distortion-dilation effect [5]. In the present talk, this ongoing efforts of our group, being in progress by piecing together their reported various $a_0(ss)$, local-structure, $\sigma(ion)$ and thermodynamic *etc* data with our own latest ¹⁵⁵Gd-Mössbauer, EXAFS and Y-MAS-NMR *etc* data obtained for several of them, will be described and discussed, in pursuit for the more global understanding of the whole DF oxides, either parent F- or stabilized P-type.

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